

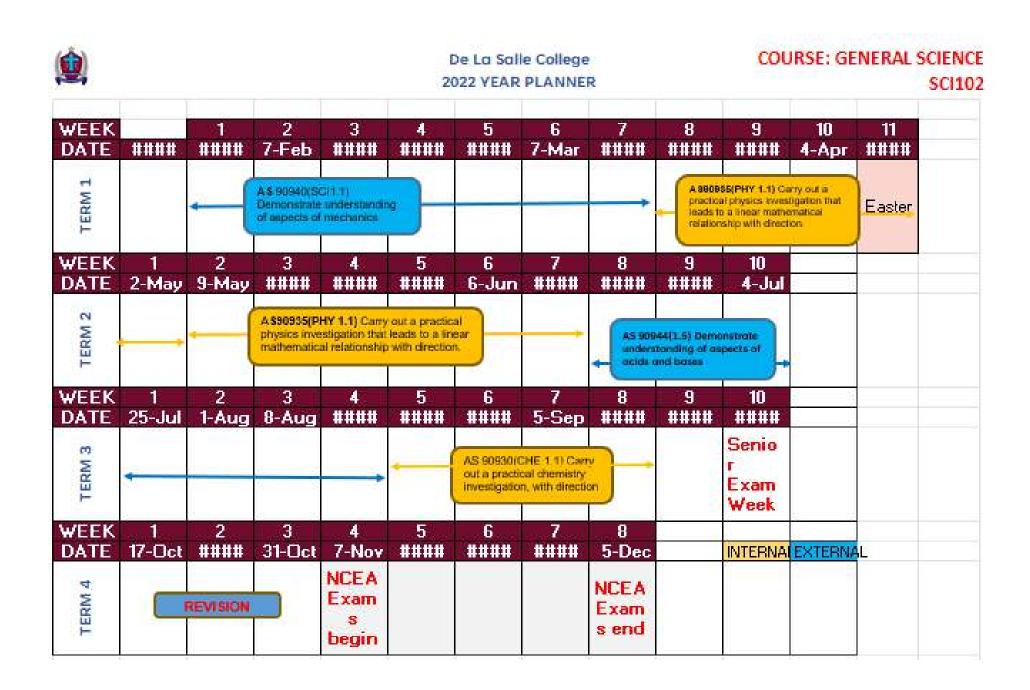
Inspiring the imagination and seeking new heights

DE LA SALLE COLLEGE STUDENT HANDBOOK 2022 Learning Area / Subject: GENERAL SCIENCE

SCI102 Year Level: 11

Curriculum Levels: 6-7

NCEA LEVEL One



De	La	Sal	le	Col	lege
20.		~		~ ~ ~	

Wed, 24 Nov 2021



SCI102 Assessment Statement 2022

Course is endorsable

Year :11 Course : Science	MR A KUMAR	Total Credits : 20
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Level 1 Science is compulsory in Year 11. This course is a full Science Course based on Level 6 of the New Zealand Curriculum. This course includes 2 external and 3 internal standards worth a total of 20 credits. Students will develop a broad range of Scientific Skills and Knowledge. Topics include: Acids and Bases, Mechanics and Life Processes of Microbes. Students will be expected to complete practical investigation and write scientific report on their findings. Upon successful completion of this course, students will begin to appreciate the role of Science in society, and will be able to continue Science specialised courses (Chemistry, Biology, Physics and Earth and Space) in Year 12.

Pre Requisites

A satisfactory pass in Year 10 Science assessment and examination

Additional Requirements Student workbook cost: \$20 for Year 11 SciPad.

Scientific Calculator

No	Standard Number	Version	Level	Credits	Lit / Num	Full Title	Method of Assessment	Assessment Opportunities Offered	Approximate Date	Grade	Teacher Signature
1	90935	3	1	4	Num	Physics 1.1 - Carry out a practical physics investigation that leads to a linear mathematical relationship, with direction	Practical	1	Week 3 Term 2		
2	90950	3	1	4		Science 1.11 - Investigate biological ideas relating to interactions between humans and micro-organisms	Assignment	1	Week 9Term 2		
3	90930	3	1	4	Num	Chemistry 1.1 - Carry out a practical chemistry investigation, with direction	Practical	1	Week 10Term 3		
4	<mark>90940</mark>	3	1	4	Num	Science 1.1 - Demonstrate understanding of aspects of mechanics	Exam	External	Term 4		
5	90944	4	1	4		Science 1.5 - Demonstrate understanding of aspects of acids and bases	Exam	External	Term 4		
6						Prelim Exam for: Science 1.1 - Demonstrate understanding of aspects of mechanics	Exam		Week 10 Term 3		

School Assessment Procedures - You can view your rights and obligations in the school's assessment procedures in the Student Assessment Handbook

Record your internal grades and ask your teacher to sign it off as correct. You can then use this as evidence of your achievement.

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Science

2022 Course Outline - Y11 General Science (SCI102)

Subject: General Science NCEA Level: One

Entry Requirements: General Science skills from Y10

Number of credits gained: 20

Method of assessment:

- Both internal and external assessment
- Internal assessments are given after a series of mock practice runs
- Mock externals at the end of each unit of work
- Mid Year Examinations
- Preliminary Examinations

Looking Ahead:

- Level 2 and 3 Biology, Chemistry, Physics and Earth & Space Science
- Tertiary level study
- A diverse range of careers stem from Biology, Chemistry, Physics and the Earth Sciences:

e.g. biosecurity officer, pharmacist, chemical and mechanical engineering, GP, electrician, ecologist, climatologist (this is just a fraction of what is possible!).

Course Description

Course aims:

This course is aimed at all general ability students entering from Year 10 and who may wish to pursue a career that requires chemistry and/or physics / biology / earth science. Such careers include engineering, medicine, architecture, pilot and more.

The course consists of units from the separate Level One Chemistry and Science subjects and provides a sound preparation for Level 2 Physics, Chemistry, Biology and Earth and Space courses.

The Science component covers: mechanics, acids and bases, genetics. The Chemistry component covers: chemical reactions.

Course learning outcomes:

- To be able to carry out a practical investigation, with direction, by planning the investigation, collecting and processing the data, and interpreting and reporting the findings.
- To be able to describe chemical reactions, including the carrying out of calculations.
- To be able to describe the atomic structure and bonding found in a variety of substances (metallic, ionic and molecular)
- To be able to describe the characteristic properties and reactions of metals, acids and bases.
- To develop knowledge and understanding of mechanics in one dimension and the use of appropriate methods to solve related problems.
- To develop fundamental knowledge and understanding of genetics and the inheritance of traits.

2022 Course Assessment Statement – Y11 General Science (SCI102)

HOW WILL I BE ASSESSED IN THIS SUBJECT?

Achievement Standard	Level and Credit Value	Internal or External Assessment	Brief Description	My grade for prelims	My final grade for internals
PHY 1.1 AS 90935 (INT) Carry out a practical physics investigation that leads to a linear mathematical relationship with direction.	Level 1 4 Credits	INTERNAL	Students will run a practical investigation to investigate a linear relationship between two physical quantities. They will follow a scientific method, collect data, plot graph to determine a mathematical relationship.		
CHEM AS1.1 (90930) Carry out a practical chemistry investigation, with direction	Level 1 4 Credits	INTERNAL	Using rates of reaction, students will conduct an investigation into how concentration can affect how fast or slow a chemical reaction takes to proceed.		
SCI1.11 AS 90950 Investigate biological ideas relating to interactions between humans and micro-organism.	Level 1 4 Credits	INTERNAL	Students will investigate how a bacteria, fungi or virus affects humans. They will investigate the personal and social implications. The investigation could be supported with a practical.		
SCI AS1.1 (90940) Demonstrate understanding of aspects of mechanics	Level 1 4 Credits	EXTERNAL	In this standard, students will learn about speed, acceleration, forces, energy and pressure. Mechanics is the scientific study of movement.		
SCI AS1.5 (90944) Demonstrate understanding of aspects of acids and bases	Level 1 4 Credits	EXTERNAL	Students will develop an understanding around atomic structure, formulae and ions. Also included is information on the chemistry behind acids and bases.		

Note: Depending on class ability. Extra internal assessments might be added if required and if the challenge and time is warranted.



De La Salle College - General Science SCI102 - Year Planner 2022

Subject: 11 General Science

Teacher in charge: Mr A KUMAR

Year Level: 11 (

Curriculum Levels: 6

Unit Title: AS90940 (AS1.1 Science External) - Demonstrate Unit Title: AS90950 (AS1.11 Biology Internal) Investigate biological **Unit Title:** AS90935 (AS1.1 Physics Internal) – Carry out a ideas relating to interactions between humans and micro-organism practical physics investigation that leads to a linear mathematical understanding of aspects of mechanics. Achievement objectives: relationship, with direction. Achievement objectives: Achievement objectives: • Understand that scientists' investigations are informed by current scientific theories and aim to collect evidence that will Demonstrate understanding of aspects of mechanics • Investigate trends and relationships in physical phenomena be interpreted through processes of logical argument. providing evidence that shows awareness of how simple (in the areas of mechanics, electricity, electromagnetism, facets of phenomena, concepts or principles relate to Use a wider range of science vocabulary, symbols, and heat, light and waves, and atomic and nuclear physics). given situations. This may include using methods for conventions. Understand that scientists' investigations are informed by solving problems involving aspects of mechanics. current scientific theories and aim to collect evidence that Apply their understandings of science to evaluate both popular • Demonstrate in-depth understanding of aspects of will be interpreted through processes of logical argument. and scientific texts (including visual and numerical literacy). mechanics providing evidence that shows how or why Relate key structural features and functions to the life phenomena, concepts or principles relate to given Learning outcomes/skills: processes of plants, animals, and micro-organisms and situations. investigate environmental factors that affect these processes. Demonstrate comprehensive understanding of aspects of A practical investigation is an activity that includes collecting, *mechanics* providing evidence that shows how or why processing and interpreting data. The investigation must lead Learning outcomes/skills: phenomena, concepts and principles are connected in the to a linear mathematical relationship context of given situations. Statements must demonstrate understanding of connections between concepts. To develop a report that arises from a investigation which involves Assessment tasks/method: collecting information about interactions between humans and micro-organisms. The information could come from a variety of Learning outcomes/skills: sources such as direct observations, collection of experimental Students will learn the steps of a scientific investigation and data, resource sheets, photos, videos, websites, and reference carry out a directed practical investigation to collect data, To develop knowledge and understanding of mechanics • texts. process data, analyze and draw a conclusion. Carry out a and the use of appropriate methods to solve related practical physics investigation involves: problems. Assessment tasks/method: • developing a method for collecting the data • collecting primary data, with units, relevant to the purpose, Students will carry out an investigation whereby biological ideas relating to how humans use and are affected by micro-organisms. based on the manipulation of the independent variable over a reasonable range and number of values Assessment tasks/method: Micro-organisms will be selected from: bacteria, fungi and viruses. Students will observations or findings to describe/explain how humans controlling the variable(s) that could have a significant effect on the results use or are affected by micro-organisms. Mid Year Mock Examinations (Term 2). ٠ significant links about the interactions between humans and micro-• using technique(s) that increase the accuracy of the organisms, including the impacts of this knowledge on human's personal measured values of the dependent (and independent, if Preliminary Mock Examinations (Term 3). actions or everyday life. It may involve explaining, elaborating, applying, appropriate) variable justifying, relating, evaluating, comparing and contrasting, and analysing. • drawing a graph, based on the data External NCEA examination at year's end. . • writing a conclusion that links the processed data to the Kev competencies: Thinking, Managing self, Using language. identified trend on the graph. Key competencies: Thinking, Managing self, Using language symbols, and texts, Participating and contributing. symbols, and texts, Participating and contributing. Values: Inquiry and curiosity, Excellence. Key competencies: Thinking, Managing self, Using Values: Inquiry and curiosity, Excellence. language, symbols, and texts, Participating and contributing. Approximate time required: 5 weeks Approximate time required: 6 weeks Values: Inquiry and curiosity, Excellence. Approximate time required: 6 weeks

Unit Title: AS90930 (AS1.1 Chemistry Internal) - Carry out a	Unit Title: AS90944 (AS1.5 Science External) – Demonstrate an	
practical chemistry investigation, with direction.	understanding of aspects of acids and bases.	
Achievement objectives:	Achievement objectives:	
 Students will carry out a practical chemistry investigation which will include collecting, processing, and interpreting primary data to reach a conclusion in a chemistry context using chemistry vocabulary, symbols, conventions and equations as appropriate. The investigation can take different contexts and could include: acid-metal reactions, acids and bases, rates of reaction, energy output of fuels, fermentation. 	 Students will describe, identify, name, draw, or giving an account of aspects of acids and bases. Students will be required to use a chemistry vocabulary, symbols and conventions (including names and formulae), and completing word equations and symbol equations. Learning outcomes/skills: 	
Learning outcomes/skills:	 Students must be able to show a fundamental understanding of the following aspects of acids and bases: 	
 To be able to carry out a practical investigation, with direction, by planning the investigation, collecting and processing the data, and interpreting and reporting the findings. 	 Atomic structure electron arrangement of atoms and monatomic ions of the first 20 elements(a periodic table will be provided) isotopes ionic bonding 	
Assessment tasks/method:	 names and formulae of ionic compounds using a given table of ions. Properties acide release bydrogen ions in water 	
 developing a procedure for collecting primary data, with units, relevant to the purpose, based on the manipulation of the independent variable over a valid range of values with repetition to show reliability controlling the variable(s) that could have a significant effect on the results using techniques to increase the accuracy of the measured values of the dependent (and independent) variable processing and representing the data to enable a conclusion to be reached writing a conclusion based on the processed data that links to the purpose of the investigation. 	 acids release hydrogen ions in water reactions (of acids with bases) to form salts. pH and effects on indicators. Rates of reaction and particle theory. Uses Neutralisation carbon dioxide formation salt formation. Acids and bases are restricted to HCl, H2SO4, HNO3, metal oxides, hydroxides, carbonates and hydrogen carbonates. Assessment tasks/method: Mid Year Mock Examinations (Term 2). 	
Key competencies: Thinking, Managing self, Using language, symbols, and texts, Participating and contributing.	 Preliminary Mock Examinations (Term 3). External NCEA examination at year's end. 	
Values: Inquiry and curiosity, Excellence. Approximate time required: 7 weeks	• External NCEA examination at year's end. Key competencies: Thinking, Managing self, Using language, symbols, and texts, Participating and contributing.	
	Values: Inquiry and curiosity, Excellence.	
	Approximate time required: 7 weeks	

Number AS90940 Version

Page 1 of 2

Achievement Standard

3

Subject R	eference	Science 1	.1		
Title		Demonstr	ate unders	tanding of aspects of mecha	anics
Level	1	Credits	4	Assessment	External
Subfield	Science				
Domain	Science -	Core			
Status		Registere	d	Status date	30 November 2010
Planned r	eview date	31 Decem	ber 2020	Date version published	20 November 2014

This achievement standard involves demonstrating understanding of aspects of mechanics and may include using methods when solving related problems.

Achievement Criteria

Achievement	Achievement with Merit	Achievement with Excellence
 Demonstrate understanding of aspects of mechanics. 	Demonstrate in-depth understanding of aspects of mechanics.	 Demonstrate comprehensive understanding of aspects of mechanics.

Explanatory Notes

1 This achievement standard is derived from The New Zealand Curriculum, Learning Media, Ministry of Education, 2007, Level 6. It is aligned with the Physical Inquiry and Physics Concepts achievement objectives in the Physical World strand and the Communicating in Science achievement objective in the Nature of Science strand, and is related to the material in the Teaching and Learning Guide for Science, Ministry of Education, 2010 at http://seniorsecondary.tki.org.nz.

This standard is also derived from Te Marautanga o Aotearoa. For details of Te Marautanga o Aotearoa achievement objectives to which this standard relates, see the <u>Papa Whakaako</u>.

2 Demonstrate understanding of aspects of mechanics typically involves providing evidence that shows awareness of how simple facets of phenomena, concepts or principles relate to given situations. This may include using methods for solving problems involving aspects of mechanics.

Numbe	AS90940	Version	3	Page 2 of
pr				chanics typically involves a, concepts or principles relate
inv	olves providing evi	dence that show ted in the contex	s how or why ph t of given situation	s of mechanics typically enomena, concepts and ons. Statements must oncepts.
5 Ev	idence may be writ	ten, mathematic	al, graphical or o	diagrammatic.
6 As • •	acceleration and d journeys, sport, get Mass, weight and forces, in the contr constant speed, are Force and pressur $P = \frac{F}{A}$. Work and power, g conservation of me	nterpretation of of leceleration in the etting going. The the acceleration ext of everyday e ccelerating. The e in the context of gravitational pote echanical energy as sports perform	distance and spe e context of ever relationships v due to gravity, b experiences such relationship Fne of everyday exper- ential energy, kin v in free fall situa mance, dropping	red time graphs, average ryday experiences such as = $\frac{\Delta d}{\Delta t} = \frac{\Delta v}{\Delta t}$. ealanced and unbalanced in as being stationary, moving at t = ma. eriences. The relationship etic energy, and the tions in the context of everyday things, tossing balls. The
				ard can be accessed through .nz/ncea/resources.

Quality Assurance

- Providers and Industry Training Organisations must have been granted consent to assess by NZQA before they can register credits from assessment against achievement standards.
- 2 Organisations with consent to assess and Industry Training Organisations assessing against achievement standards must engage with the moderation system that applies to those achievement standards.

Consent and Moderation Requirements (CMR) reference 0233

Number	AS90944	Version	4		Page 1 of 3
		Achievem	ent Standar	d	
Subject R	eference	Science 1.5			
Title		Demonstrate under	standing of as	pects of acids a	and bases
Level	1	Credits	4	Assessment	External
Subfield	Science				
Domain	Science - (Core			
Status		Registered	Status date	e	30 November 2010
Planned re	eview date	31 December 2020	Date version	on published	20 November 2014

This achievement standard involves demonstrating understanding of atomic structure, particle theory and rates of reaction relating to acids and base properties, uses and reactions.

Achievement Criteria

Achievement	Achievement with Merit	Achievement with Excellence			
 Demonstrate understanding of aspects of acids and bases. 	 Demonstrate in-depth understanding of aspects of acids and bases. 	 Demonstrate comprehensive understanding of aspects of acids and bases. 			

Explanatory Notes

Version 2 of this achievement standard was republished to correct an error in the bullet points about properties of acids and bases in explanatory note 5.

This standard is also derived from Te Marautanga o Aotearoa. For details of Te Marautanga o Aotearoa achievement objectives to which this standard relates, see the Papa Whakaako.

- 1 This achievement standard is derived from The New Zealand Curriculum, Learning Media, Ministry of Education, 2007, Level 6. It is aligned with the Material World strand, and is related to the material in the Teaching and Learning Guide for Science, Ministry of Education, 2010 at http://seniorsecondary.tki.org.nz.
- Demonstrate understanding typically involves describing, identifying, naming, 2 drawing, or giving an account of aspects of acids and bases. This may require the use of chemistry vocabulary, symbols and conventions (including names and formulae), and completing word equations.

- Number AS90944 Version 4 Page 2 of 3 3 Demonstrate in-depth understanding typically involves explaining aspects of acids and bases. This may require explanations that use chemistry vocabulary, symbols and conventions (including names and formulae) and writing word equations or completing given symbol equations. Demonstrate comprehensive understanding typically involves linking aspects of acids 4 and bases. It may involve explaining, elaborating, justifying, relating, evaluating, comparing and contrasting, or analysing. This may require the use of chemistry vocabulary, symbols and conventions (including names and formulae), and writing balanced symbol equations. 5 Aspects of acids and bases will be selected from: Atomic structure - electron arrangement of atoms and monatomic ions of the first 20 elements (a periodic table will be provided) ionic bonding names and formulae of ionic compounds using a given table of ions. Properties acids release hydrogen ions in water - reactions (of acids with bases) to form salts - pH and effects on indicators. Rates of reaction and particle theory. Uses - neutralisation carbon dioxide formation - salt formation. Acids and bases are restricted to HCI, H₂SO₄, HNO₃, metal oxides, hydroxides, 6 carbonates and hydrogen carbonates. Other acids may be included in examination questions. The names and formulae of any such acids will be given in the question.
 - 7 Assessment Specifications for this achievement standard can be accessed through the Science Resources page found at www.nzga.govt.nz/ncea/resources.

Number	AS90935	Version	3		Page 1 of 3	Nu	mber	AS90935	Version	3	Page 2 of 3
		Achieveme	ent Standa	rd		2					<i>Guidance Manual for New Zealand</i> 2000, must be followed during the
Subject R	eference	Physics 1.1						cal investigation.		11.11.11.11.174.17	
Title				ics investigatio onship, with dir	on that leads to a rection	3	writing	g and direction w	ill be given in the	form of a	the investigation will be specified in purpose, an outline of the method,
Level	1	Credits	4	Assessment	t Internal						o choose. A template or suitable led for the student to use.
Subfield	Science					4	A pra	ctical investigatio	on is an activity th	at includes	s collecting, processing and
Domain	Physics										a linear mathematical relationship.
Status		Registered	Status dat	Ð	30 November 2010	5	-	out a practical p			s:
Planned r	eview date	31 December 2020	Date versi	on published	20 November 2014		• co ma		data, with units, r	elevant to	the purpose, based on the a reasonable range and number of

This achievement standard involves carrying out a practical physics investigation that requires the graphical representation and mathematical description of a linear relationship, with direction.

Achievement Criteria

Achievement	Achievement with Merit	Achievement with Excellence		
• Carry out a practical physics investigation, with direction, that leads to a linear mathematical relationship.	 Carry out an in-depth practical physics investigation, with direction, that leads to a linear mathematical relationship. 	Carry out a comprehensive practical physics investigation, with direction, that leads to a linear mathematical relationship.		

Explanatory Notes

1 This achievement standard is derived from *The New Zealand Curriculum*, Learning Media, Ministry of Education, 2007, Level 6. It is aligned with the Physical Inquiry and Physics Concepts achievement objective in the Physical World strand and the Investigating in Science achievement objectives in the Nature of Science strand, and is related to the material in the *Teaching and Learning Guide for Physics*, Ministry of Education, 2010 at <u>http://seniorsecondary.tki.org.nz</u>.

This standard is also derived from Te Marautanga o Aotearoa. For details of Te Marautanga o Aotearoa achievement objectives to which this standard relates, see the Papa Whakaako.

- writing a conclusion that links the processed data to the identified trend on the graph.
- 6 Carry out an in-depth practical physics investigation involves:
 - · controlling the variable(s) that could have a significant effect on the results
 - using technique(s) that increase the accuracy of the measured values of the dependent (and independent, if appropriate) variable
 - drawing a linear graph, valid for the data

drawing a graph, based on the data

- writing a conclusion that states the equation of the relationship.
- 7 Carry out a comprehensive practical physics investigation involves writing a discussion that validates the conclusion. The discussion may include as appropriate:
 - a justification for the accuracy-improving techniques used
 - a reason that there is a limit to either end of the value chosen for the independent variable
 - a justification why a variable needs to be controlled.
 - a description of any difficulties encountered when making measurements and how these difficulties were overcome
 - a link between investigation findings and applicable physics ideas
 - a description of any unexpected outcomes of the processing of the results and a suggestion of how these outcomes could have been caused and/or the effect they had on the validity of the conclusion.
- 8 Conditions of Assessment related to this achievement standard can be found at <u>http://ncea.tki.org.nz/Resources-for-Internally-Assessed-Achievement-Standards.</u>

Number AS90950 Version

6 7

Achievement Standard

3

Subject Reference Science 1.11

Title		Investigate biological ideas relating to interactions between humans and micro-organisms			
Level	1	Credits	4	Assessment	Internal
Subfield	Science				
Domain	Science - C	Core			
Status		Registered	Status date	,	30 November 2010
Planned re	view date	31 December 2020	Date versio	n published	20 November 2014

This achievement standard involves investigating biological ideas relating to how humans use and are affected by micro-organisms.

Mutual exclusion exists between this standard and AS90927.

Achievement Criteria

Achievement	Achievement with Merit	Achievement with Excellence
 Investigate biological ideas relating to interactions between humans and micro-organisms. 	 Investigate, in depth, biological ideas relating to interactions between humans and micro- organisms. 	 Investigate, comprehensively, biological ideas relating to interactions between humans and micro- organisms.

Explanatory Notes

1 This achievement standard is derived from *The New Zealand Curriculum*, Learning Media, Ministry of Education, 2007, Level 6. It is aligned with the Life Processes achievement objective in the Living World strand and the Investigating in Science, Communicating in Science, and the Participating and Contributing achievement objectives in the Nature of Science strand, and is related to the material in the *Teaching and Learning Guide for Science*, Ministry of Education, 2010 at http://seniorsecondary.tki.org.nz.

This standard is also derived from Te Marautanga o Aotearoa. For details of Te Marautanga o Aotearoa achievement objectives to which this standard relates, see the <u>Papa Whakaako</u>.

Number	AS90950	Version	3	Page 2 of 3
and direc	micro-organisms.	The information	could come from a	ractions between humans variety of sources such as ce sheets, photos, videos,
Sch		dia, Ministry of E		e Manual for New Zealand t be followed during any
100	The second s	a second second second		NY AND CONTRACTOR OF CONTRACTOR

- 3 *Investigate* involves using observations or findings to describe how humans use or are affected by micro-organisms.
- 4 *Investigate in depth* involves using findings and biological ideas to explain how or why humans use or are affected by micro-organisms.
- 5 Investigate comprehensively involves using findings and biological ideas to make significant links about the interactions between humans and micro-organisms, including the impacts of this knowledge on human's personal actions or everyday life.
 It may involve explaining, elaborating, applying, justifying, relating, evaluating, comparing and contrasting, and analysing.

Micro-organisms will be selected from: bacteria, fungi and viruses.

- The ways that humans use and are affected by micro-organisms may include: disposal of organic wastes, sewage treatment, composting, food production and preservation, food poisoning, microbial action on everyday materials (helpful and harmful micro-organisms), disease in humans and animals they are in contact with, antibiotics, resistance to antibiotics, and origins and control of pandemics.
- 8 *Biological ideas* relating to how humans use and are affected by micro-organisms may include the following:
 - structure and life processes of micro-organisms
 - culturing of micro-organisms
 - · factors that affect the life processes of micro-organisms.
- 9 Conditions of Assessment related to this achievement standard can be found at <u>http://ncea.tki.org.nz/Resources-for-Internally-Assessed-Achievement-Standards.</u>

Number	AS90930	Version	3	Page 1 of 3	Number	AS90930	Version	3	Page 2 of 3
		Achieveme	ent Standard						oose. A template or suitable r the student to use.
Subject R	eference	Chemistry 1	.1		4 A p	ractical chemistry i	investigation incl	udes collecting.	processing, and interpreting
Title		Carry out a p	practical chemistry investiga	ation, with direction	prin	nary data to reach	a conclusion in a	a chemistry conte	ext using chemistry oppropriate. Suitable contexts
Level	1	Credits	4 Assessmen	t Internal	cou		etal reactions, ad		ates of reaction, energy output
Subfield	Science				UT I	delo, termentation.			
Domain	Chemistry					ry out a practical c developing a meth			h units, relevant to the
Status		Registered	Status date	30 November 2010		purpose, based on values	the manipulatio	n of the independ	dent variable over a range of
Planned r	eview date	31 December 2020	Date version published	20 November 2014	- 2	processing and rep calculation etc) writing a conclusio			iate way (graph, table,

This achievement standard involves carrying out a procedure to collect and process primary data and interpret the results, with direction.

Achievement Criteria

Achievement	Achievement with Merit	Achievement with Excellence
 Carry out a practical chemistry investigation, with direction. 	 Carry out an in-depth practical chemistry investigation, with direction. 	 Carry out a comprehensive practical chemistry investigation, with direction.

Explanatory Notes

1 This achievement standard is derived from *The New Zealand Curriculum*, Learning Media, Ministry of Education, 2007, Level 6. It is aligned with the Nature of Science and Material World strands and is related to the material in the *Teaching and Learning Guide for Chemistry*, Ministry of Education, 2010 at http://seniorsecondary.tki.org.nz.

This standard is also derived from Te Marautanga o Aotearoa. For details of Te Marautanga o Aotearoa achievement objectives to which this standard relates, see the Papa Whakaako.

- 2 The procedures outlined in Safety and Science: A Guidance Manual for New Zealand Schools, Learning Media, Ministry of Education, 2000, must be followed during the practical investigation.
- 3 With direction means that general instructions for the investigation will be specified in writing and direction will be given in the form of a purpose, an outline of the method,

- 6 Carry out an in-depth practical chemistry investigation involves:
 - developing a procedure for collecting primary data, with units, relevant to the purpose, based on the manipulation of the independent variable over a valid range of values with repetition to show reliability
 - · controlling the variable(s) that could have a significant effect on the results
 - using techniques to increase the accuracy of the measured values of the dependent (and independent) variable
 - · processing and representing the data to enable a conclusion to be reached
 - writing a conclusion based on the processed data that links to the purpose of the investigation.
- 7 Carry out a comprehensive practical chemistry investigation involves an in-depth investigation that also:
 - justifies the choices made to increase accuracy during the investigation
 - justifies the conclusion in terms of the processed data and the purpose of the investigation
 - · relates investigation findings to applicable chemistry ideas.
- 8 Conditions of Assessment related to this achievement standard can be found at <u>http://ncea.tki.org.nz/Resources-for-Internally-Assessed-Achievement-Standards.</u>

Replacement Information

This achievement standard replaced AS90169.

2022 General Science (SCI102) – Student Guide to Bibliographies / Referencing

A bibliography is the 'trail' of reading that you did to inform your thinking for your essay or assignment. A bibliography is organised alphabetically by the author's last name.

Setting up a bibliography

(a) Books

- Author's last name and initial
- Date published in brackets
- Name of book underlined
- Place of publication
- Publisher

Single Author

Eg. Ward, M. (1995) Celebrating Women Cambridge, University Press

More than one author

Eg Edwards, B., Horrocks, N. (1992) Reporting for Television Palmerston North, Dunmore Press

- (b) Journals, Magazines, Newspapers
 - Author's name and initial
 - Date published in brackets
 - Title of the article in quotation marks
 - Name of paper or magazine –underlined
 - Place of publication
 - Volume (if applicable)
 - Page number

Eg Booth, P. (1986) "The Arthur Thomas Case Revisited" NZ Herald Feb 11, 2: 13

- (c) Internet
 - Author's name and initial if available
 - Article name if appropriate
 - Full date of resource if available
 - Date of access
 - Site address

Eg Midbon, M (2000), Jan 13 2004 <u>A Day Without Yesterday: George Lemaitre and the Big Bang</u> <u>http://catholiceducation.org/articles/science/sc0022.html</u>

To create automatically correct references, the following website is useful. It will create references for ISBN numbers and webpages.

www.harvardgenerator.com

	e La Salle College 2022 sment Result Appeal Form
Name:	Class:
Name/number of standard being appealed	l:
Subject:	Teacher who marked work:
Grade awarded for standard:	
Date work returned to student:	Date of appeal:
Reason for appeal:	
Student signature:	Caregiver's signature:
OFFICE USE ONLY Teacher response:	
HOF response:	
Principal's Nominee response:	
Final decision:	

De L	a Salle College 2022
Absence F	From Internal Assessment
Appli	ication for Extension
Student:	Class:
Subject:	Teacher:
Assessment title:	
Standard number:	
Type of assessment activity (test, practical, a	assignment etc).
Date of assessment or due date:	
Reason for application:	
Illness or injury: <i>medical certificate of</i>	r a letter from parent / caregiver must be attached.
Family / personal trauma: documenta counsellor or Dean).	tion must be attached (eg. a letter from parent / caregiver,
School activity (sporting or cultural) _	
	used to determine a grade. HOD / TIC to attach
	etermine the grade and the grade awarded.
The reason for this has been explained to me	e and I accept the decision.
Signed: (Studen	